

SOV/112-59-2-2326

Propagation of Electromagnetic Waves Over a Periodic Structure of Conducting . . . .  
determined by superposing plane waves whose amplitudes are periodic  
functions of two coordinates. The dispersion equation set up in the article is  
solved graphically. Plots are presented of propagation and attenuation  
constants against the wavelength-to-structure-period ratio and also against  
the plate resistance per unit area.

V. Ye.B.

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"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

KOLPAKOV, V.V.

Propagation of electromagnetic waves in a periodic structure of  
conducting plates. Izv.vys.ucheb.zav.; radiofiz. 1 no.2:2-7 '58  
1. Sibirskiy fiziko-tehnicheskiy institut pri Tomskom universitete.  
(Radio waves)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

KOLPAKOV, V.V.

Reflection of plane electromagnetic waves from a periodic structure  
of conduction plates. Izv.vys.ucheb.zav.; radiofiz. 1 no.2:8-12 '58.  
1. Sibirskiy fiziko-tehnicheskiy institut pri Tomskom universitete.  
(Radio waves)

9.1912

S/194/62/000/005/108/157  
D230/D308

AUTHOR: Kolpakov, V.V.

TITLE: Reflection of plane electromagnetic waves from a semi-limited periodic structure consisting of conducting plates

PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika, no. 5, 1962, 24, abstract 5zh177 (Tr. Sibirs. fiz.-tekhn. in-ta pri Tomskom un-te, 1960, no. 39, 25-33)

TEXT: A study of the reflection of a plane monochromatic e.m. wave from a semi-restricted periodic structure consisting of thin plates with finite conductivity. The problem is solved using approximate boundary conditions for thin plates. [Abstractor's note: Complete translation].

Card 1/1

9,1300

29311  
S/109/61/006/010/009/027  
D201/D302AUTHOR: Kolpakov, V.V.

TITLE: Diffraction of surface electromagnetic waves at the impedance discontinuity of a circular cylinder

PERIODICAL: Radiotekhnika i elektronika, v. 6, no. 10, 1961,  
1658 - 1664

TEXT: The author analyzes the problem of diffraction of electromagnetic surface waves at the impedance discontinuity of an infinite impedance cylinder. Since this cylindrical system finds application in surface wave propagation systems, the solution of this problem is both of technical and practical interest. The problem consists of integrating the differential equation of an axially symmetric TM-wave for the sole component of its magnetic field with boundary conditions

$$E_z = ZH_\phi \quad (1)$$

at the surface of the impedance cylinder and with conditions at in-

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Diffractio<sup>n</sup> of surface ...

finity. In the plane  $z = 0$

$$Z = \begin{cases} Z_1 = -iQ_1 & \text{for } z < 0 \\ Z_2 = -iQ_2 & \text{for } z > 0 \end{cases} \quad (2)$$

and thus with  $k = \omega/c = 2\pi/\lambda$  the inhomogeneous Wiener-Hopf integral equation

$$H(a, z) = H_1^{(1)}(av_1)e^{ih_1 z} + 2\pi ak(Q_2 - Q_1) \int_0^\infty G(a, a, z, z') H(a, z') dz' \quad (9)$$

is obtained which is solved in the complex variable plane by the usual methods and results in

$$H(p, z) = H_1^{(1)}(pv_1)e^{ihz} + \frac{k(Q_2 - Q_1)}{2\pi i} H_1^{(1)}(av_1) \Psi_1(h_1) \int_{-\infty}^{+\infty} e^{-i\omega z} \frac{H_1^{(1)}(\rho v) d\omega}{(\omega + h_1)\Psi_2(\omega)[vH_0^{(1)}(av) + kQ_2 H_1^{(1)}(av)]}. \quad (18)$$

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which can be further processed and gives

$$H(\rho, z) = H_1^{(1)}(\rho v_1) e^{ih_1 z} + \frac{h_1 - h_2}{h_1 + h_2} \exp M(h_1) H_1^{(1)}(\rho v_1) e^{-ih_1 z}, \quad z < 0, \quad (19a)$$

the first term of which represents an incident at the discontinuity surface wave, the second term representing the reflected wave. The field of the transmitted surface wave is derived as

$$H(\rho, z) = \frac{2h_1}{h_1 + h_2} \sqrt{\frac{|H_1^{(1)}(av_1)|^2 - H_0^{(1)}(av_1) H_2^{(1)}(av_1)}{|H_1^{(1)}(av_2)|^2 - H_0^{(1)}(av_2) H_2^{(1)}(av_2)}} \times \\ \times \exp \left[ \frac{M(h_1)}{2} - \frac{M(h_2)}{2} \right] H_1^{(1)}(\rho v_2) e^{ih_2 z}, \quad z > 0. \quad (19b)$$

The power reflection coefficient R and transmission coefficient T are determined by

$$R = \left( \frac{\beta_1 - \beta_2}{\beta_1 + \beta_2} \right)^2 \exp 2X(\beta_1), \quad (20)$$

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$$T = \frac{4\beta_1\beta_2}{(\beta_1 + \beta_2)} \exp [x(\beta_1) - x(\beta_2)], \quad (20)$$

in which  $\beta_1 = h_1/k$  and  $\beta_2 = h_2/k$  - the delay factors at both sides of the impedance discontinuity. From the analysis of graphs of relevant expressions it is shown that  $R$ ,  $T$  and  $\eta$  depend on  $a/\lambda$  only for small values of  $a/\lambda$ . For  $a/\lambda > 1$ ,  $R/R_\infty$ ,  $T/T_\infty$  and  $\eta/\eta_\infty$  differ little from unity and for the above values of  $a/\lambda$  the coefficients  $R$ ,  $T$  and  $\eta$  may be evaluated from the approximate formulae

$$R = \frac{\beta_1^2 - 1}{\beta_1^2(\beta_1 + \beta_2)^2} (\sqrt{\beta_1^2 - 1} - \sqrt{\beta_2^2 - 1})^2 \exp \left( \frac{3}{4a^2k^2} \frac{\sqrt{\beta_2^2 - 1} - \sqrt{\beta_1^2 - 1}}{(\beta_1^2 - 1)\sqrt{\beta_2^2 - 1}} \right), \quad (22)$$

$$T = \frac{4\sqrt{\beta_1^2 - 1}\sqrt{\beta_2^2 - 1}}{(\sqrt{\beta_1^2 - 1} + \sqrt{\beta_2^2 - 1})^2} \exp \left[ \frac{3}{8a^2k^2} \frac{(\sqrt{\beta_2^2 - 1} - \sqrt{\beta_1^2 - 1})^2}{(\beta_1^2 - 1)(\beta_2^2 - 1)} \right].$$

For  $a/\lambda \rightarrow 0$ , the function  $X(u)$  is zero and coefficients  $R$ ,  $T$  and

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tend to

$$R_0 = \left( \frac{\beta_1 - \beta_2}{\beta_1 + \beta_2} \right)^2, \quad T_0 = \frac{4\beta_1 \beta_2}{(\beta_1 + \beta_2)^2}, \quad \eta_0 = 0;$$

the radiation field resulting from the diffraction of a surface wave at the impedance discontinuity is determined by going over to the new variable  $\tau$ :  $w = k \cos \tau$ ,  $v = k \sin \tau$ , and to the spherical system of coordinates and by evaluating the integral of Eq. (18) for large values of  $k\tau$  (saddle point  $\tau = \pi - \theta$ ). Thus

$$\begin{aligned} H(r, \theta) &= \frac{1}{\pi i} \frac{e^{ikr}}{r} \times \\ &\times \sqrt{\frac{2(Q_2 - Q_1) \frac{h_1^2(h_1 - h_2)}{k^2(h_1 + h_2)} ak \{ [H_1^{(1)}(av_1)]^2 - H_0^{(1)}(av_1) H_2^{(1)}(av_1) \} \exp\left[\frac{M(h_1)}{2}\right]}{V[\sin \theta H_0^{(1)}(ak \sin \theta) + Q_1 H_1^{(1)}(ak \sin \theta)] [\sin \theta H_0^{(1)}(ak \sin \theta) + Q_2 H_1^{(1)}(ak \sin \theta)]}} \times \\ &\times \frac{k(k \cos \theta + h_2)}{\sqrt{(k^2 \cos^2 \theta - h_1^2)(k^2 \cos^2 \theta - h_2^2)}} \exp\left[-\frac{M(k \cos \theta)}{2}\right]. \end{aligned} \quad (23)$$

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D201/D302

is found. The solution for a TE-wave is not given since all formulae as obtained for a TM-wave remain valid for a TE-wave, provided  $H_\varphi$  is substituted by  $E_\varphi$ ;  $E_\psi$  by  $-H_\rho$ ;  $E_z$  by  $-H_z$  and  $Z$  by  $1/Z$ . There are 5 figures and 5 references: 3 Soviet-bloc and 2 non-Soviet-bloc. The reference to the English-language publication reads as follows: A.F. Kay, Scattering of a surface wave by a discontinuity on reactance, IRE, IRE, Trans. 1959, AP-7, 1, 22.

SUBMITTED: January 18, 1961

Card 6/6

9.1912

8/194/62/000/007/105/160  
D271/D308

AUTHOR: Kolpakov, V.V.

TITLE: Approximate boundary conditions for conducting plate

PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika,  
no. 7, 1962, abstract 7zh208 (Tr. Sibirs. fiz.-tekhn.  
in-ta pri Tomskom un-te, 1960, no. 39, 18-24)

TEXT: Approximate boundary relations between tangential components  
of electric and magnetic fields are derived for an infinite plate  
with a finite conductance. In a particular case when plate thick-  
ness is smaller than that of the skin layer, the approximate boun-  
dary conditions assume an especially simple form. The problem of  
reflection of a plane wave from a conducting film is considered as  
an example of application of the approximate boundary conditions.  
[Abstracter's note: Complete translation.]

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$$\text{FWC}(t)/\text{EWP}(e)/\text{EPA}(e) - 2/\text{EWT}(m)/\text{EPR}(e) \times \text{EWT}(t)/\text{EPA}(e) = (\text{EPR}/\text{EPA}(w) - 2/T)/\text{EWP}(t)/\text{EWP}(w)$$

AR5015168 1F/0137/63,000/005/0038/0038

zh. Metallurgiya, Abs. 50226

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Чубин, Б. Т.; Лебедева, Л. П.; Игнатьев, И. Н.; Чепаков, В. В.

data on the technology of manufacturing articles of friction  
with material, brand FMK-11 [Translator's note: original gives FMK-P-7]

Tr. I Vses. nauchno-tehn. konferentsii po zerkach. metallurgii.

total ceramic material, friction material, total mechanical and physical property, iron, powder metal, and polymer, metal ceramic.

In investigation has been made of the effect of the technological conditions of production on the mechanical and friction properties of the material. The use of oxidized powders is recommended. An increase in the oxygen content of the powder is influenced by the wear of the material and the connected parts (Chilean cast iron).

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Ref: AR5015168

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use in pressing pressure, the wear of the material and the cast iron  
The optimum pressing pressure is 6 to 8, cm<sup>3</sup>. Use of continuous  
~~favoring conditions permitted shortening the sintering time to 4.5 hrs and~~  
the technology of the process. Sintering was done under a pressure  
of 10 kg/cm<sup>2</sup>. To avoid overheating of disks, it is recommended to use  
the paint (22-24% black graphite, 74-76% quartz sand, 2-3%  
clay, remainder water). The paint is applied to the packed disks  
sintering without special preparation of the surface. V. Kras.

SUB CODE: MM

ENCL: 00

AP501787  
UR, 0266, 65/0700, 1/0118/0119  
621, 325  
*71*  
*B*

Stepchenko, I. M.; Krysin, B. T.; Kolpakov, Yu. V.; Smirnov,  
Kolyubyskiy, V. A.; Taytaenko, M. V.; Vetrov, N. P.; Vinogradov,  
I. I.; Levin, M. M.; Edel'man, M. I.

Method for producing friction parts from powder components.  
No. 171702

Pravilaten' izotreteleniy i tevarnykh znamen, No. 1, 1955,

Aircraft brake, friction part, : sinter metallurgy,  
Author Certificate has been issued for the method of pro-  
duction parts (e.g., brake-unit parts) of aircraft from pow-  
der. To reduce wear, the mixture consists of 20-30% iron,  
3-10% barium sulfate, 3-7% silica, 10-12% asbestos,  
and sintere at a temperature of 1200°C and a pressure of  
[LB]

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TO: none

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ENCL: 00

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OTHER: 000

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L 00598-66 EMP(a)/EMP(b)-2/SNT(m)/EPF(c)/EMP(1)/ENA(d)/EPA(w)-2/T/EMP(t)/EMP(k)/  
EMP(z)/EMP(b) ER/ID/WN/JG/DJ/WN

ACCESSION NR: AR5018950

URL/0276/65/000/007/B047/B047  
611.762.3:621.81

SOURCE: Ref. zh. Tekhnologiya mashinostroyeniya. Svednyy tom, Abs. 7B333

48

B

AUTHOR: Kravtsov, B. T., Lebedeva, I. P., Ismatov, L. N., Kolpakov, Ya. V.

TITLE: New developments in the technology of part manufacture from FM-11 friction cermets

CITED SOURCE: Tr. VII Vses. nauchno-tekhn. konferentsii po peroshk. metallurgii,  
Yerevan, 1964, 257-263

TOPIC TAGS: powder metallurgy, friction cermet, oxidized metal powder, process factor effect, antiwear compound, FM-11 cermet

TRANSLATION: The authors studied the effects of original material condition and sintering-compacting/process factors on the physico-mechanical and friction properties of FMK-11 cermet. It is shown that use of oxidized Fe and Cu powders simplifies the technological process (i.e. eliminates reduction and crushing operations), reduces the manhours required and decreases the cost of a finished piece without lowering quality. The cermet is characterized by high strength and hardness, as well by minimal wear, when compacted at a pressure of about 6 t/cm<sup>2</sup>.

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L 00598-66

ACCESSION NR: AR5018950

An effective pressure of 20 - 25 kg/cm<sup>2</sup> is optimal for the sintering process. The sintering period was reduced from 7 to 4.5 - 5.0 hours when temperature was increased steadily to assigned levels. The period required to preheat the workpiece to an assigned temperature was reduced from 4.5 to 2.0 - 2.5 hours. The authors recommend the use of an antiscorch coating composed of 22 - 24% black graphite, 14 - 16% quartz sand, 9 - 10% refractory clay and the balance in water. Use of the coating facilitates automation of the coating application process and an improved industrial hygiene environment. Two illustrations and 7 tables.

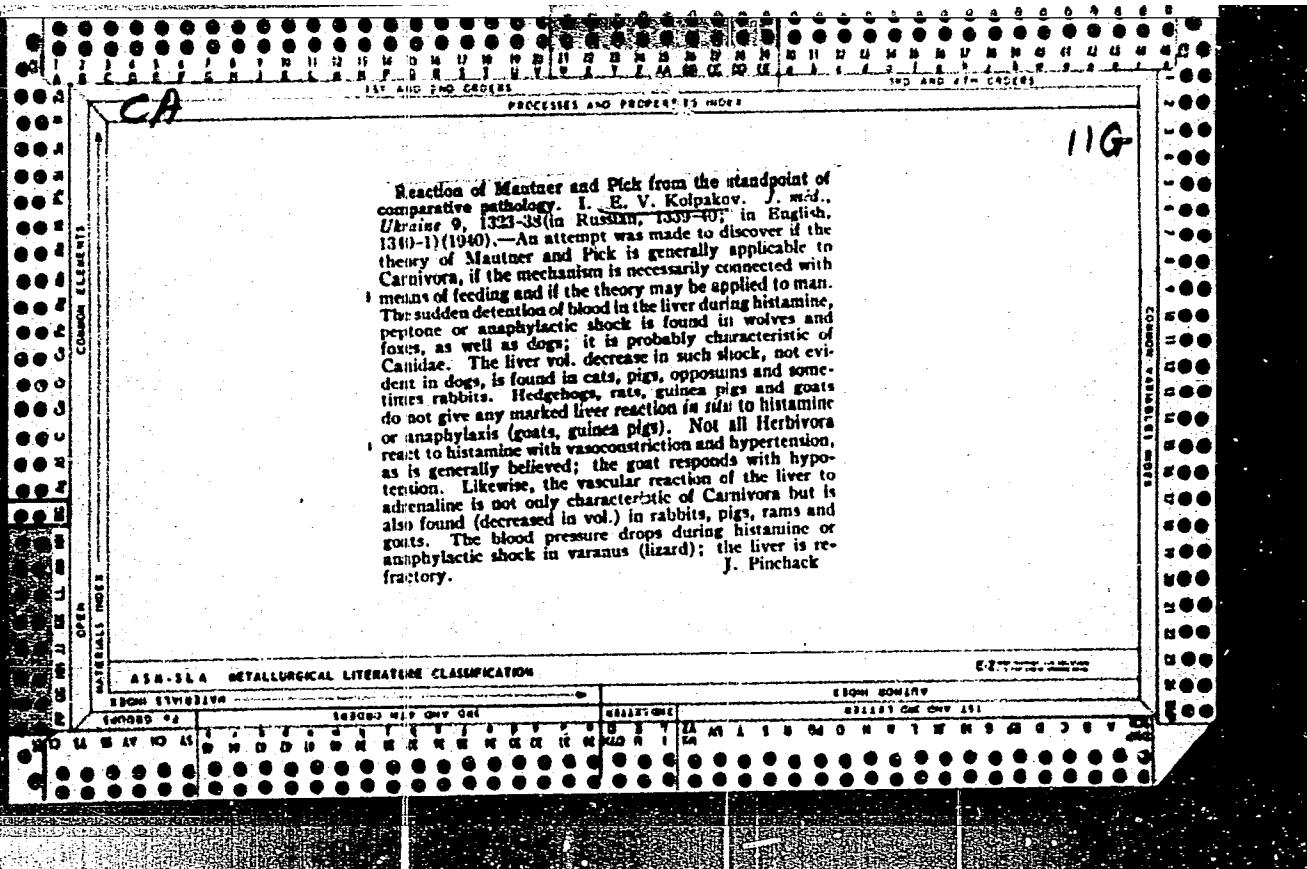
SUB CODE: MF

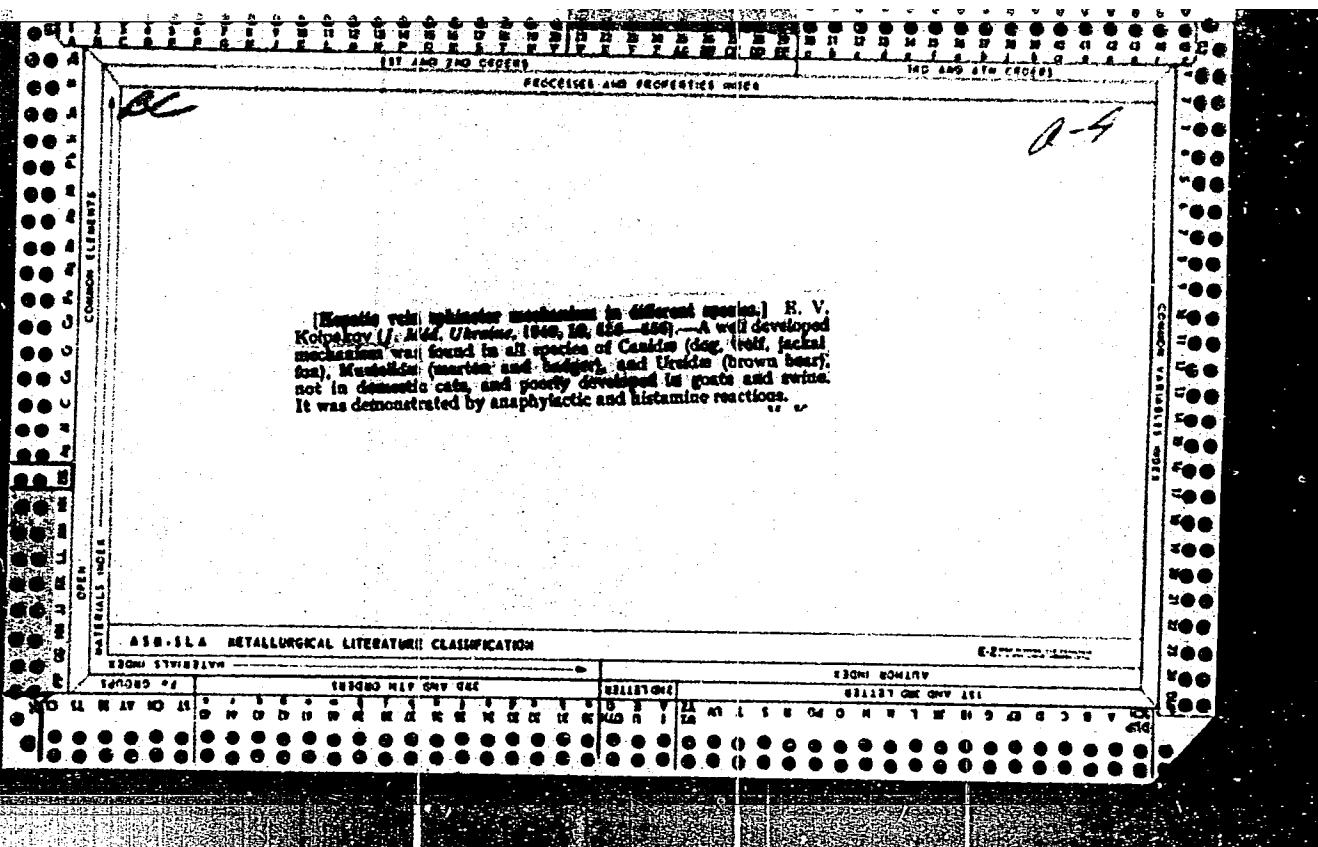
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Card: 1/2

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Kolpakov, Ye.N.

KOLPAKOV, Ye.V., doktor biol.nauk

Comparative study of muscular sphincters of the hepatic veins.  
Medich.zhur. 19 no.2:14-30 '49. (MIRA 10:12)

1. Z viddilu porivnyal'noi patologii (zav. viddilu - chlen-kor. AN URSR M.M.Sirotinin) Instituta klinichnoi fiziologii im. akad. O.O.Bogomol'tsya AN URSR (direktor - chlen-kor. AN URSR P.Ye.Kavets'-kiy).

(HEPATIC VEINS)

KOLPAKOV, YE. V.

25957. Kolpakov, Ye. V. Sravnitel'nye issledovaniye myshechnykh shlyuzov pechenchnykh von. Med. zhurnal, T. xix, Vyp. 2, 1954, s. 14-30.---Na ukr. yaz.---Rezyume na rus. yaz.---Bibliogr: 16 nazv.

SO: Knizhnaya Letopis', Vol. 1, 1955

KOLPAKOV, Ye.V.; RAYETS'KA, H.P.

Erroneous views on the mechanism which regulates blood circulation in the liver. Medich. zhur. 22 no.6:73-82 '52.  
(MLRA 6:10)

1. Instytut klinichnoyi fiziologiyi im. O.O. Bohomol'tsya Akademiiyi nauk UESR.  
(Liver) (Blood--Circulation)

KOLPAKOV, Ye. V.; LAUER, N.V.

Effect of experimental hepatic insufficiency on conditioned reflex activity in dogs. Vop. fiziol. no.10:44-57 '54 (MLRA 10:5)

1. Institut fiziologii im. A.A. Bogomol'tsa Akademii nauk USSR.  
(LIVER--DISEASES) (CONDITIONED RESPONSE)

АСЛІМАЧ, Я. В.

CHAGOVETS, Vasiliy Yuri'yevich; BABSKIY, Ye.B., akademik, otvetstvennyy redaktor; KAVETSKIY, R.Ye., akademik, redaktor; KOLPAKOV, V., professor, redaktor; MAKARENKO, A.F., redaktor; VOL'BOBT, Yu.V., akademik, redaktor; SHENZHEN, M.I., redaktor izdatel'stva; KOLOMIYCHUK, V.A., tekhnicheskiy redaktor.

[Selected works; in one volume] Izbrannye trudy; v odnom tome. Kiev, Izd-vo Akad.nauk USSR, 1957. 513 p. (MIRA 10:11)

1. Akademiya nauk USSR (for Babskiy, Kavetskiy, Vol'bort).
2. Chlen-korrespondent Akademii nauk USSR (for Makarchenko).  
(Electrophysiology)

KOLPAKOV, Ye.V.

Nikolay Nikolayevich Sirotinin; on his 60th birthday. *Fiziol.zhmr.*  
[Ukr.] 3 no.1:12-15 Ja-F '57. (MLRA 10:3)  
(SIROTIININ, NIKOLAI NIKOLAEVICH, 1896- )

KOLPAKOV, Ye.V. [Kolpakov, YE.V.]

Vascular reflexes of the liver as related to the action of the sphincters of liver veins. Fiziol.zhur. [Ukr.] 5 no.3:322-328  
My-Je '59. (MIRA 12:10)

1. Institut fiziologii im. O.O.Bogomol'tsaya AN URSR, laborato-  
riya porivnyal'noi i vikovoi fiziologii.  
(LIVER--BLOOD VESSELS)

Согласовано

KOLPAKOV, Ye.V. [Kolpakov, YE.V.]

Charles Darwin; on the 150th anniversary of his birth. Fiziol.  
zhur. [Ukr.] 5 no.3:295-300 My-Je '59. (MIRA 12:10)  
(DARWIN, CHARLES ROBERT, 1809-1882)

KOLPAKOV, Ye.V. [Kolpakov, И.В.]

On the methodology used in creating an Ekk-Pavlov fistula.  
Fiziol. zhur. 6 no.1:125-130 Ja.-F '60. (MIRA 13:5)

1. Institut fiziologii im. A.A. Bogomol'tsa AN USSR.  
(FISTULA)

KOLPAKOV, Ye.V. [Kolpakov, I.E.V.]

Comparative physiology of blood circulation in the liver. Fiziol.  
zhur. [Ukr.] 7 no.3:395-408 My-Je '61. (MIRA 14:5)

1. Laboratoriya sravnitel'noy i vozrastnoy fiziologii Instituta  
fiziologii im. A.A.Bogomol'tsa AN USSR, Kiyev.  
(LIVER—BLOOD SUPPLY)

MAKARCHENKO, A.F., akademik, otv. red.; SIROTININ, N.N., zam. otv. red.; KOLPAKOV, Ye.V., prof., red.; LAUER, N.V., doktor med. nauk, red.; GUREVICH, M.I., doktor med. nauk, red.; KOLCHINSKAYA, A.Z., kand. med. nauk, red.; YANKOVSKAYA, Z.B., red. izd-va; BEREZOVSAYA, D.N., tekhn. red.

"Oxygen deficiency; hypoxia and adaptation to it] Kislorod-naya nedostatochnost'; gipoksiia i adaptatsiia k nei. Kiev, Izd-vo AN USSR, 1963. 609 p. (MIRA 17:2)

1. Akademiya nauk URSR, Kiev. Instytut fiziologii. 2. Akademiya nauk Ukr. SSR (for Makarchenko). 3. Deystvitel'nyy chlen AMN SSSR (for Sirotinin).

KOLPAKOV, Ye.V. [Kolpakov, IE.V.]

Volodymyr Ivanovich Vernads'kii; on the 100th anniversary of his  
birth. Fiziol.zhur.[Ukr.] 9 no.1:3-5 Ja-F '63.

(MIRA 18:5)

LEBEDEVA, L.P.; KRYGIN, B.T.; KOLPAKOV, Ya.V.; IGNATOV, L.N.;  
MIKHAYLOVSKIY, V.A.; SMIRNOV, G.G.; TSYTSENKO, M.V.

Experimental production of iron-base friction ceramic metals.  
Porosh. met. 5 no.8:96-102 Ag '65. (MIRA 18:9)

GOLYSHEV, A.B., kand. tekhn. nauk; POLISHCHUK, V.P., inzh.; KOLPAKOV, Yu.A.,  
inzh.

Solving a relaxation problem during the calculation of continuous  
combined structures for the settling of supports. Sbor. trud. Inzh.-  
stroi. fak. Chel. politekh. inst. no.3:31~41 '63. (MRA 17:9)

1. Ural'skiy filial Akademii stroitel'stva i arkhitektury SSSR.

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BERSENEV, A.P., kand.tekhn.nauk; KOLPAKOV, Yu.D., inzh.

Determining the moisture content of wood by its dielectric permeability. Der.prom. 9 no.12:13+14 D '60.  
(MIRA 13:12)  
(Wood-Moisture)

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CIA-RDP86-00513R000824010006-0"

S/081/61/000/024/010/086  
B138/B102

AUTHORS: Skripov, V. P., Kolpakov, Yu. D.

TITLE: Scattering of light in carbonic acid along sub- and trans-critical isotherms

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 24, 1961, 73, abstract 24B522 (Sb. "Kritich. yavleniya i flyuktuatsii v rastvorakh". M., AN SSSR, 1960, 126 - 136)

TEXT: The scattering of light on carbonic acid is studied for five sub- (19.98 - 30.67°C) and five transcritical (31.20 - 49.90°C) isotherms. Intensity of scattered I' and of transmitted light I are measured for three lines of the mercury spectrum (5461, 4350 and 4060 Å). The measurements were made with high pressures (50 - 125 at) in the system. The extremum values of I' increase on approaching critical temperature, and the difference between I' for the liquid and the vapor becomes less. On the transcritical isotherms I' peaks are observed, which also increase on approaching critical point. With variable p - t, points for the I' peaks of transcritical isotherms plot very well into a straight line, merging

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Scattering of light in carbonic...

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with the saturated vapor pressure curve at critical point. The coordinates of the critical point are  $31.00^{\circ}\text{C}$  and  $73.2 \pm 0.5$  atm. The results are in good agreement with the Rayleigh scattering law  $I' = 1/\lambda^4$ , with the exception of the isotherm peak at  $31.20^{\circ}\text{C}$ , which is closer to critical point, where  $I' = 1/\lambda^{3.2}$ . [Abstracter's note: Complete translation.] ✓

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KOLPAKOV, Yu.D.

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PHASE I BOOK EXPLOITATION SOV/5469

Sovoshehniye po kriticheskim yavleniyam i flyuktuatsiyam v rastvorakh. Moscow, 1960.

Kriticheskiye yavleniya i flyuktuatsii v rastvorakh; trudy sovashchniya, yanvar' 1960 g. (Critical Phenomena and Fluctuations in Solutions; Transactions of the Conference, January 1960) Moscow, Izd-vo AN SSSR, 1960. 190 p. 2,500 copies printed.

Sponsoring Agencies: Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk. Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova. Khimicheskly fakul'tet.

Responsible Ed.: M. I. Shakhpuronov, Doctor of Chemical Sciences, Professor; Ed. of Publishing House: E. S. Dragunov; Tech. Ed.: S. G. Tikhomirova.

PURPOSE : This collection of articles is intended for scientific personnel concerned with chemistry, physics, and heat power engineering.

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**Critical Phenomena and Fluctuations**

SOV/5469

COVERAGE: The book contains 24 of the 26 reports read at the Conference on Critical Phenomena and Fluctuations in Solutions organized by the Chemical Division of Moscow State University, January 26-28, 1960. The reports contain results of investigations carried out in recent years by Soviet physicists, chemists, and heat power engineers. The Organizing Committee of the Conference was composed of Professor Kh. I. Amirkhanov, A. Z. Golik, I. R. Krichevskiy (Chairman), V. K. Semenchenko, A. V. Storonkin, I. Z. Fisher, and M. I. Shakharonov (Deputy Chairman). References accompany individual articles.

## TABLE OF CONTENTS:

## Foreword

3

Amirkhanov, Kh. I., A. M. Kerimov, and B. G. Alibekov [Laboratoriya molekulyarnoy fiziki, Dagestanskiy filial AN SSSR -- Laboratory of Molecular Physics, Dagestan Branch, AS USSR]. Thermophysical Properties of Matter at Critical Temperature

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## Critical Phenomena and Fluctuations

SOV/5469

Alhadov, Ya. Yu., and M. I. Shakhparonov [Laboratoriya fiziko-khimii rastvorov, Khimicheskiy fakul'tet, Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova -- Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Dielectric Properties of Solutions in a Superhigh Frequency Field at Concentration Fluctuations

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Beridze, D. K., and M. I. Shakhparonov [Laboratory of Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Light Scattering in Solutions Having a Critical Stratification Point

21

Vuks, M. F., and L. I. Lisnyanskiy [Laboratoriya molekulyarnoy optiki, Fizicheskiy fakul'tet, Leningradskiy gosudarstvennyy universitet im. A. A. Zhdanova -- Laboratory of Molecular Optics, Physics Division, Leningrad State University imeni A. A. Zhianov]. Intermolecular Interaction and Light Scattering in Solutions of Pyridine and  $\alpha$ -Picoline in Water

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Zatsepina, L. P., and M. I. Shakhparonov [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Rayleigh Light Scattering in Nitrobenzene -- Cyclohexane and Ethyl Alcohol - Diethylamine Solutions

32

Kasimov, R. M., and M. I. Shakhparonov [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Dielectric Properties of Solutions in Electromagnetic Fields of the Millimetric Band and Concentration Fluctuations

37

Krichevskiy, I. R., and N. Ye. Khazanova [Laboratoriya vysokikh davleniy. GIAP -- Laboratory of High-Pressure [Studies], Moscow State Design and Planning Scientific Research Institute of the Nitrogen Industry]. Diffusion of Liquid and Gaseous Solutions in the Critical Region

45

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## Critical Phenomena and Fluctuations

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High-Pressure [Studies], GIAP]. Kinetics of Heterogeneous Processes in the Critical Region 54

Krichevskiy, I. R., N. Ye. Khazarova, and L. R. Linshits [Laboratory of High Pressure [Studies], GIAP]. Liquid-Vapor Equilibrium in the Critical Region of Liquid-System Stratification 61

Lomova, N. N., and M. I. Shakharonov [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Permittivity and Molecular Structure of Solutions 73

Lanshina, L. V., and M. I. Shakharonov [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Thin Structure of the Line of Rayleigh Light Scattering in Solutions 77

Mokhov, N. V., and Ya. M. Labkovskiy [Kafedra eksperimental'noy fiziki, Dnepropetrovskiy gosudarstvennyy universitet -- Depart-

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30

ment of Experimental Physics, Dnepropetrovsk State University].  
Investigation of Density Fluctuations in Ether and Benzene  
Based on X-Ray Scattering at Narrow Angles

81

Mol'khov, N. V., and I. V. Kirsh [Department of Experimental  
Physics, Dnepropetrovsk State University] Variation in the  
Sizes of Concentration Fluctuations in Relationship to Tem-  
perature and Concentration in Binary Liquid Systems Having  
an Upper Critical Dissolving Temperature

89

Nozdrev, V. F., B. I. Kal'yanov and M. G. Shirkevich [Moskov-  
skiy oblastnoy pedagogicheskiy institut -- Pedagogical Insti-  
tute of the Moscow Oblast]. Hypersonic Investigation in  
Organic Liquids at Constant Density in the Vicinity of the  
Critical State

93

Rott, L. A. [Minskiy lesotekhnicheskiy institut -- Minsk  
Forestry Engineering Institute]. Concerning the Diffusion in  
the Critical Stratification Region

102

Card 6/9

30

Critical Phenomena and Fluctuations

SOV/5469

Roshchina, G. P. [Laboratoriya molekulyarnoy fiziki, Fizicheskiy fakul'tet, Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko -- Laboratory of Molecular Physics, Division of Physics, Kiev State University imeni T. G. Shevchenko] Investigation of Fluctuations in Solutions by the Method of Light Scattering

109

Skripov, V. P. [Laboratoriya molekulyarnoy fiziki, Ural'skiy politekhnicheskii institut im. S. M. Kirova -- Laboratory of Molecular Physics, Ural Polytechnic Institute imeni S. M. Kirov]. Special Structural Features of Matter in the Vicinity of the Critical Point and Transfer Phenomena

117

Skripov, V. P., and Yu. D. Kolpakov [Laboratory of Molecular Physics, Ural Polytechnic Institute imeni S. M. Kirov, and the Laboratoriya teplofiziki, Ural'skiy filial AN SSSR -- Thermophysics Laboratory, Ural Branch, AS USSR]. Light Scattering in Carbon Dioxide along Pre- and Post-Critical Isotherms

126

Smirnov, B. A. [Institut neftekhimicheskogo sinteza AN SSSR -- Card 7/9

30

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Fisher, I. Z. [Belorussky Gosudarstvennyy Universitet -- Belorussian State University (Minsk)] Correlation Analysis of the Critical Point	148
Shakhparonov, M. I. [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Fluctuations in Solutions	151
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Card 8/9

30

Critical Phenomena and Fluctuations SOV/5469

Shimanskaya, Ye. T., Yu. I. Shimanskiy, and A. Z. Golik [Laboratory of Molecular Physics, Division of Physics, Kiyev State University imeni T. G. Shevchenko]. Investigation of the Critical State of Pure Substances by Topler's Method 171

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AVAILABLE: Library of Congress (QD545.S73)

JP/dfk/jw  
10-28-61

Card 9/9

5.4100

80284  
S/170/60/003/04/05/027  
B007/B102

AUTHORS: Skripov, V.P., Kolpakov, Yu.D.

TITLE: An Investigation of the Interphase-region Transition in Carbonic Acid From Light Scattering

PERIODICAL: Inzhenerno-fizicheskiy zhurnal, 1960, Vol. 3, No. 4, pp. 30-36

TEXT: In this paper experiments are described in which scattered as well as transmitted light was observed during isothermal change of the state of the substance. The experiments were made with carbonic acid. A section through the test chamber is shown in Fig. 1. The experimental arrangement consisted of this chamber, of a system for refilling the device with carbonic acid, a thermal pressure control and an optical system. The device is briefly described. Eight isothermal curves (6 transcritical and 2 subcritical) of the intensity of scattered and of transmitted light as depending on carbonic-acid pressure were taken. The entire temperature range of the measurements was 80°C. The dependence of height and position of the maxima of  $I'$  (intensity of scattered light) on the magnitude of the difference  $\Delta T$  between testing temperature and critical temperature is very conspicuous in these curves (Fig. 2). The intensity rise of

Card 1/3

80284

An Investigation of the Interphase-region Transition  
in Carbonic Acid From Light Scattering

S/170/60/003/04/05/027  
B007/B102

scattered light becomes weaker with growing distance from  $T_c$  (critical temperature). Vertical lines mark the points of condensation in the below-critical isothermal lines. Fig. 3 shows the dependence of the  $I'$ -maxima on  $\Delta T$  for three spectral lines. The maximum of light scattering shifts towards higher pressure with rising temperature (Fig. 2). The relation between temperature and pressure at the  $I'$ -maxima is, near the critical point, a straight line with the inclination of  $dp/dl' = 1.50$  at/deg or, in reduced quantities,  $dx/dr = 6.2$ . The latter value is almost equal to that obtained by M.G. Kaganer (Ref. 6) for the critical isochoric curve of various nonpolar gases ( $dx/dr = 6.0$ ). In the experiments described also the intensity  $I$  of the transmitted light was measured. The minima of transmittent light were obtained in the range of the scattering maxima. The results of earlier measurements made by one of the authors (V.P. Skripov) and G.P. Nikolayev (Ref. 11) have already been given. The qualitative dependence of light scatter on wavelength is shown in table 1 and Fig. 3. The light scattering observed had the character of a Rayleigh scattering, i.e.  $I' \sim 1/\lambda^4$ . The data of light scatter and formula (3) (Ref. 8) may be used for calculating elasticity  $\gamma = -(dp/dv)_T$  or compressibility  $\beta = \frac{1}{v} (dv/dp)_T$  of the substance

Card 2/3

KOLPANCY, Yu.D.; SKRIPOV, V.P.; GORBUNOVA, E.N.

Scattering of light in carbonic acid and its relation to the  
equation of state. Ukr.fiz.zhur. 7 no.7:787-792 Jl '62.

(MIRA 12:15)

1. Ural'skiy politekhnicheskiy institut i Ural'skiy filial AN  
SSSR, g. Sverdlovsk.

(Light—Scattering) (Carbonic acid) (Equation of state)

L. 11997-66 EWT(1)/EWT(m)/ETC(m) IJP(c)/RPL WW/JW/GG/RM

ACC NR: AP022863

SOURCE CODE: UR/0051/65/019/003/0392/0402

52

B

AUTHOR: Skripov, V. P.; Kolpakov, Yu. D.

ORG: none

44,55

21,49,55

TITLE: Light scattering in the vicinity of the critical liquid-vapor point. I. Apparatus. Experiments with carbon dioxide and sulfur hexafluoride

SOURCE: Optika i spektroskopiya, v. 19, no. 3, 1965, 392-402

TOPIC TAGS: carbon dioxide, sulfur compound, light scattering, phase transition, critical point

ABSTRACT: This is a continuation of earlier work by the authors (Ukr. fiz. zh. v. 7, 787, 1962 and earlier), where particular attention was paid to the connection between scattered light and the nature of supercritical transitions in carbon dioxide. In the present investigation improved equipment was used, and greater attention was paid to methods of investigating opalescence. In addition, measurements were made for the first time on sulfur hexafluoride. The light scattering was determined from the isotherms<sup>144,55</sup> in a broad region near the critical points of CO<sub>2</sub> and SF<sub>6</sub> using a pressure-regulated thermostatic chamber. The lines 5780, 5461, 4358, and 4046 Å from the mercury spectrum were used as sources, and the transmitted (unpolarized) and scattered light was recorded photoelectrically. The apparatus and procedure are described in detail. The critical parameters of the carbon dioxide and sulfur hexafluoride were found to be 31.06 and 45.55°C (critical temperature) and 73.1 and 37.7 atm (critical

Card 1/2

UDC: 535.36

L 11997-66  
ACC NR: AP5022863

pressure), respectively. These agree well with data by others. The scattering abilities of CO<sub>2</sub> and SF<sub>6</sub> are compared and the ratio of the maximum intensity of the scattered light of SF<sub>6</sub> is found to exceed that of CO<sub>2</sub> by a factor 1.77, which compares well with the calculated average value 1.83. Orig. art. has: 7 figures, 4 formulas, and 6 tables.

SUB CODE: 20/ SUBM DATE: 10Jun64/ ORIG REF: 010/ OTH REF: 005

Card 2/2

KOLPAKOV, YU. G.

Kolpakov, YU. G. "Track automobile roads of constant type  
(Investigation of the operation of the thoroughfare portion  
of automobile roads)." Min Higher Education Ukrainian SSR.  
Kiev Automobile and Road Inst. Kiev, 1956. (Dissertations  
for the Degree of Candidate in Technical Science)

So: Knizhnaya letopis', No. 27, 1956. Moscow. Pages 94-109; ill.

KOLPAKOV, Yu.G., kand.tekhn.nauk

Using granite chippings from stone quarries for bituminous pavements. Avt.dor.i dor.stroi. no.l:185-192 '65.

(MIRA 18:11)

6513 69648

S/078/60/005/05/27/037  
B004/B016

5.2200(A)

AUTHORS: Filinov, F. M. (Deceased), Tekster, Ye. N., Kolpakova, A. A.,  
Panteleyeva, Ye. P.

TITLE: Investigation of the Solubility of Thorium Pyrophosphate in Acids,  
and Investigation of the Equilibrium Between Solid Phase and  
Solution in the Systems  $\text{ThP}_2\text{O}_7$  -  $\text{Na}_4\text{P}_2\text{O}_7$  -  $\text{H}_2\text{O}$  and  $\text{ThP}_2\text{O}_7$  -  $\text{Th}(\text{NO}_3)_4$ -  
 $\text{H}_2\text{O}$

PERIODICAL: Zhurnal neorganicheskoy khimii, 1960, Vol. 5, No. 5, pp. 1149 - 1156

TEXT: The solubility of  $\text{ThP}_2\text{O}_7$  was studied in  $\text{HCl}$ ,  $\text{HNO}_3$ , and  $\text{H}_2\text{SO}_4$  in the concentration range 0.1 - 0.5 N by means of the radiochemical indicator  $\text{UX}_1$ . The method is described in a paper by F. M. Filinov and V. F. Budanova (Ref. 1). The activity of the solutions was measured on the B-apparatus by means of a B-1  $\beta$ -counter. Fig. 1 shows the vessel applied. Data on solubility are presented in Fig. 2, and for comparison, also the data of Ref. 8 are given. The solubility of  $\text{ThP}_2\text{O}_7$  in solutions of  $\text{Na}_4\text{P}_2\text{O}_7$  in the concentrations range 0.02 - 0.2 M was also determined

Card 1/2

V. N. I. AP3000643  
APFTC, ASL, SSD

S/0000/63/236 203/0536/0536

Kil'gakova, A. A., Markovakiy, L. Ya.

34  
52

TITLE: Luminescent properties of BaO-SiO<sub>2</sub> systems activated by cerium and manganese

SOURCE: Zhurnal prikladnoy khimii, v. 36, no. 3, 1963, 530-536

TOPIC TAGS: Ba-silicates, cerium, manganese, luminescence, luminophores

ABSTRACT: An investigation of the luminescent properties of Ba silicates activated by the combined action of trivalent Ce and Mn showed luminescence dependent upon the transitions of the silicate. Ba<sub>3</sub>SiO<sub>5</sub> and Ba<sub>2</sub>SiO<sub>4</sub> are non-luminescent; Ba<sub>3</sub>SiO<sub>5</sub> has a very weak, and Ba<sub>2</sub>SiO<sub>4</sub> a strong light blue,  $\lambda = 440$ -millimicron emission; and BaSi<sub>2</sub>O<sub>5</sub> and Ba silicates richer in SiO<sub>2</sub> (BaSi<sub>2</sub>O<sub>5</sub> admixed with SiO<sub>2</sub>) give a white luminosity,  $\lambda = 400$  and  $\lambda = 500$  millimicrons. X-rays of Ce or Mn silicates were formed. In the transition phases from Ba<sub>3</sub>SiO<sub>5</sub> to BaSi<sub>2</sub>O<sub>5</sub> where x-rays showed BaSi<sub>2</sub>O<sub>5</sub> - the luminosity prevailed; this was explained by the greater intensity of Ce by traces, not detected by x-rays of Ba<sub>3</sub>SiO<sub>5</sub> - Ce, Mn, as confirmed by mechanical mixtures of the former with 5-10% of the latter

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14675-63  
ACCESSION NR: AP3000643

silicate. Extensive study was made to determine the optimum roasting temperature, effect of BaF<sub>2</sub> sub 3 and BaCl<sub>2</sub> sub 2 as flux, and concentration of activators for the preparation of these luminophores: 1100° with 3% BaCl<sub>2</sub> flux and 10% CeO<sub>2</sub> plus 1.4% Mn. The emission of the luminophore Ba sub 2 Si sub 3 O sub 8 - Ce, Mn is not more than 35% than emission from ZnS - Ag and 30% greater than from MgWO sub 4. In conclusion the authors express their appreciation to Yu. D. Kondrashev for conducting X-ray structural analyses of test specimens. Orig. art. has: 3 tables, 6 figures.

2

ASSOCIATION: Gosudarstvennyy institut prikladnoy khimii (State Institute of Applied Chemistry)

SUBMITTED: 27Jul62

DATE ACQ: 12Jun63

ENCL: 00

SUB CODE: CH

NO REF Sov: 007

OTHER: 013

Card 2/2

ACCESSION NR: AT4044998

S/3110/64/000/051/0102/0106

AUTHOR: Kolpakova, A.A., Markovskiy, L. Ya.

TITLE: Luminescent properties of the magnesium silicate system after activation with cerium and manganese

SOURCE: Leningrad. Gosudarstvennyy Institut prikladnoy khimii. Trudy\*, no. 51, 1964. Khimiya i tekhnologiya lyuminosoforov (Chemistry and technology of luminophors), 102-106

TOPIC TAGS: luminophor, magnesium silicate, activated magnesium silicate, silicate luminescence, cerium, manganese, luminescence spectrum, absorption spectrum

ABSTRACT: The phase composition of the MgO-SiO<sub>2</sub> system, which is important in the manufacture of ceramics and refractory materials, has been studied in detail, but its luminescent properties are insufficiently understood. The authors therefore studied the relationship between phase composition and luminescence in the MgO-SiO<sub>2</sub> system after activation with CeO<sub>2</sub> (5 wt. %) and/or MnCl<sub>2</sub>·4 H<sub>2</sub>O (1%) by x-ray and spectroscopic analysis of samples with varying composition (in steps of 5 mol. %). In order to achieve equilibrium, the luminophor was heated for 20 hrs. at 1200°C in the presence of MgF<sub>2</sub>

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ACCESSION NR: AT4044998

before activation. The effect of the activating elements on the luminescence and absorption spectra are shown in Fig. 1 of the Enclosure. The experimental data show that the  $Mg_2SiO_4$  phase does not show luminescence when activated with Ce alone, and produces only a very faint dark-red light in the presence of Mn. The  $MgSiO_3$  phase also produces very faint luminescence when activated with either Mn (red light) or Ce (blue light) alone, but in the presence of both Mn and Ce the luminescence shows two peaks (390 and 670 m $\mu$ ). The strongest luminescence was produced by activated lumino-phors containing excess  $SiO_2$ , which apparently facilitates the formation and activation of protoenstatite. "Yu. D. Kondrashev participated in the x-ray studies." Orig. art. has: 1 table and 2 figures.

ASSOCIATION: Gosudarstvennyy institut prikladnoy khimii, Leningrad (State Institute of Applied Chemistry)

SUBMITTED: 00

ENCL: 01

SUB CODE: IC

NO REF SOV: 007

OTHER: 010

Card 2/3

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

KOLPAKOVA, A.A.; PEKERMAN, F.M.

Luminophors with a silicate base and radiation in the ultra-violet  
band of the spectrum. [Trudy] GIPKH no.51:88-96 '64. (MIRA 18:5)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

MR: AP5017771

UR/0080/65/038/0071,432/1426  
535.37+546.42+31+340.284

Volpakova, A. A.; Markovskiy, L. Ya.

24  
22  
B

luminous properties of the SrO - silicon system induced by cerium and

ura. prikladnoy khimii, v. 28, no. 7, 1954, p. 1251.

strontium silicate, luminespher, containing cerium, fluorescence

the transformations in the SrO - silicon system and its association with the luminous properties. X-ray diffraction analysis evidence that in the  $\text{Sr}_3\text{SiO}_5$  phase does not form a new intermediate phase. The glow of the luminophor ( $\lambda_{\text{max}} = 550 \text{ m}\mu$ ) is due to the  $\text{Sr}_3\text{SiO}_5$  phase having a whitish-pink glow at the same time. Luminescence ( $\lambda_{\text{max}} = 350 \text{ m}\mu$ ). To determine the effect of the activators (cations) during the luminescence, we measured the absorption measured without the activators, with one activator at a time, two activators. It was found that cerium does not activate

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L-63056-65

ACCESSION NR: AP5017771

2

luminescence. In  $\text{SrSiO}_3$ -(e-Mn) luminescence due to cerium luminescence takes addition to the blue glow of cerium activation with  $\lambda_{\text{max}} = 400 \text{ nm}$ , addition of energy to the manganese ion leads to appearance of the appearance of a new band with  $\lambda_{\text{max}} = 580 \text{ nm}$ . The dependence of the intensity of the luminescence intensity of  $\text{Sr}_2\text{Si}_2\text{O}_5$ -(e-Mn) on the concentration of the activators and the temperature of the samples, the conditions of annealing were established. The results of the experiments are given in the annex and the author carried out the x-ray studies of the samples. There are 2 figures and 2 tables.

ORIGINATOR: Gnedarstvennyy Institut prikladnoy khimii (State Institute of Chemistry)

30 May 63

ENCL: 20

ATTN CODE: IC, OP

004

OTHER: OSA

KOLPAKOVA, A.A.; MARKOVSKIY, L.Ya.

Luminescent properties of the  $M_2O-SiO_2$  system in case of its activation by cerium and manganese. [rudy] GIPKH no.51:102-106 '64.

(MIRA 18:5)

L 16792-66 EWT(1) IJP(c)  
ACC NR: AP6002534

SOURCE CODE: UR/0286/65/000/023/0037/0037

AUTHORS: Kolpakova, A. A.; Markovskiy, L. Ya.

ORG: none

TITLE: A method for obtaining a luminophor based on strontium orthosilicate.  
Class 22, No. 176651 [announced by State Institute of Applied Chemistry  
(Gosudarstvennyy institut prikladnoy khimii)]

20  
8

SOURCE: Byulleten' izobretений и товарных знаков, no. 23, 1965, 37

TOPIC TAGS: luminophor, strontium compound, cerium

ABSTRACT: This Author Certificate presents a method for obtaining a luminophor  
based on strontium orthosilicate. The process calls for the use of an activator  
and for heating the charge. To obtain a luminophor radiating in the violet region  
of the spectrum and with a high temperature resistance of radiation, trivalent  
cerium is used as the activator.

SUB CODE: 07/ SUBM DATE: 20Mar63

UDC: 621.3.032.35:661.843

2

Card 1/15M

ACC NR: AI7007595

SOURCE CODE: UR/0104/66/000/006/0095/0096 26

AUTHOR: Chuprakov, N. M.; Borovoy, A. A.; Postnikov, N. A.; Malychev, A. A.; Magidson, E. N.; Sin'chugov, F. I.; Zoylidzon, Ye. D.; Barchaninov, G. S.; Yermolenko, V. M.; Vasil'yev, A. A.; Sokolov, N. I.; Ul'yanov, A. S.; Fedoseyev, A. M.; Sarkisov, M. A.; Rokotyan, S. S.; Azor'yev, D. I.; Arason, G. S.; Dubinskiy, L. A.; Zhulin, I. V.; Kolpikova, A. I.; Antoshin, N. N.; Krikunichik, A. B.; Kuchkin, M. D.; Preobrazhonskiy, N. Ye.; Reut, M. A.; Kheyfits, M. E.; Sharov, A. N.; Yakub, Yu. A.; Gorbunov, N. I.; Shurmukhin, V. A.; Beschinskii, A. A.

ORG: none

TITLE: Boris Sergeyovich Uspenskiy (on his 60th birthday)

SOURCE: Elektricheskiye stantsii, no. 8, 1966, 95-96

TOPIC TAGS: hydroelectric power plant, electric engineering personnel.

SUB CODE: 10

ABSTRACT: B. S. Uspenskiy was born in June 1905. He graduated from the State Electric Machine Building Institute in 1928 as an electric installation engineer. He worked in the State Electro-Technical Trust for four years, then in the All-Union ElectroTechnical Union, where he planned power construction units. Plans which he made up at that time for the electrical portion of electrical stations and sub-stations are still being used. He was involved in planning and installation of the electrical portion of hydro-electric power stations and powerful pumping stations in the Moscow-Volga Canal. During the war, he was in charge in installation of the Krasnogorskaya Heat and Electric Power Station, the planning of the Ural's Hydro-Electric Power Station and other projects. No Card 1/2

09281534

STEPANOVA, Ye.I.; KOLPAKOVA, A.S.; SOKOLOVA, G.A.

Using the phage titer growth reaction for the check of dis-infection effectiveness. Report No.1. Zhur. mikrobiol., epid. i immun. 33 no.12:107-112 D.'62 (MIRA 16:5)

1. Iz TSentral'noy kontrol'no-issledovatel'skoy laboratorii Moskovskoy gorodskoy dezinfektsionnoy stantsii.  
(DYSENTERY) (BACTERIOPHAGE)

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

KOLPAKOVA, A. Y., DEVINSKYN, M. L., and SLAVIN, G. A.

"Recovery Voltages During Fault Clearing on Long Transmission Lines With Series Capacitors", paper presented at International Conference on Large Electric Systems (CIGRE), 16th Session, Paris, 30 May-9 June 1956.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

KOLPAKOVA, A. I., BOGDANOVA, N. B., GERTSYK, A. K. , YEMELYANOV, N. P., MARKOVICH, I. M., POPKOV, V. I., SOVALOV, S. A., and SLAVIN, G. A.

Results of Some Researches, Carried out in the USSR on 600 kV long-distance Power Transmissions.

paper submitted for presentation at the Intl. Conf. on Large Electric Systems (CIGRE) 17th Biennial Session, Paris, France, 4-14 June 1958.

Electra, No. 30, Nov 57, periodical news letter issued by the CIGRE, Paris France.

ACC NR: AP6035702

(N)

SOURCE CODE: UR/0413/66/000/019/0048/0048

INVENTORS: Azovtsev, A. A.; Bolkhovitinov, V. K.; Ivanova, V. A.; Kolpakova, G. A.; Kyun, Ye. V.; Savel'yev, Yu. F.; Drozdov, A. I.; Byunau, A. E.

ORG: none

TITLE: A device for automatically controlling the movement of ship models on deeply immersed underwater vanes. Class 21, No. 186547 [announced by Central Scientific Research Institute imeni Academian A. N. Krylov (Tsentral'nyy nauchno-issledovatel'sky institut)]

SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 19, 1966, 48

TOPIC TAGS: shipbuilding engineering, model test, simulation test facility, automatic control system

ABSTRACT: This Author Certificate presents a device for automatically controlling the movement of ship models on deeply immersed underwater vanes, with the use of a tow device and of a measuring arm. The design makes it possible to accomplish the programmed changes of the model, conforming to angles of trim difference, of heeling, and of yaw. It also makes it possible to measure the instantaneous values of all these angles and the magnitudes of the vertical displacement of the model. The lower end of the measuring arm is mounted on a Cardan ball joint. The upper end of the arm is set in a control housing which is the inner frame of a second Cardan joint.

Card 1/2

UDC: 621.501.72:629.12.014.5

ACC NR: AP6035702

The outer frame of this second Cardan joint is rigidly fastened to the frame of the tow device.

SUB CODE: 13, 14/ SUBM DATE: 06Aug64

Card 2/2

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

KOLPAKOVA, L. D.

✓ Type: single and multi component. Viscous liquid and solid.

✓ Color: light yellow.

✓ Odor: strong, sharp, pungent.

✓ Appearance: crystalline or granular, finely crystallized. It is a white crystal powder.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene, chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

✓ Specific gravity: 1.25 g/cm<sup>3</sup>.

✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

✓ Specific gravity: 1.25 g/cm<sup>3</sup>.

✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

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✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

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✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

✓ Specific gravity: 1.25 g/cm<sup>3</sup>.

✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

✓ Specific gravity: 1.25 g/cm<sup>3</sup>.

✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

✓ Specific gravity: 1.25 g/cm<sup>3</sup>.

✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

✓ Melting point: 100°C.

✓ Boiling point: 180°C.

✓ Solubility: soluble in water, alcohol, ether, benzene,

chloroform, acetone, etc.

✓ Density: 1.25 g/cm<sup>3</sup>.

✓ Specific gravity: 1.25 g/cm<sup>3</sup>.

✓ Refractive index: 1.55.

✓ Optical rotation: +10°.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

LASTOVSKIY, R.P.; KOLPAKOVA, I.D.; IVANOVA, N.I.

Cyclohexylamine-N,N-diacetic acid. Met. poluch. khim.  
reak. i prepar. no.6:60-62 '62.

Benzylamine-N,N-acetoacetic acid. Ibid.:62-63

1. Vsesoyuznyy nauchno-issledovatel'skiy institut khimicheskikh  
reaktivov i osobo chistiykh khimicheskikh veshchestv.  
(MIRA 17:5)

LASTOVSKIY, R.P., doktor khimich.nauk; KOLPAKOVA, I.D., kand.khimich, nauk;  
DYATOLOVA, N.M., kand.khimich.nauk; TEMKINA, V.Ya., kand.khimich.  
nauk

Use of complexons in analytical chemistry. Zhur.VKHO 9 no. 2:  
138-145 '64.

(MIRA 17:9)

LAS TOVSKIY, R.P.; KOLPAKOVA, I.D.; MIRONOVA, Ye.I.

Benzhydrylamine-N,N-diacetic acid. Met. poluch. khim.  
reak. i prepar. no.6:63-65 '62. (MIRA 17;5)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut khimicheskikh  
reaktivov i osobu chistykh khimicheskikh veshchestv.

LASTOVSKIY, R.P.; KOLPAKOVA, I.D.; KOZHELENKO, L.I.

Aniline-N-N-diacetic-o-arsionic acid. Met. poluch. khim.  
reak. i prepar. no.6:65-67 '62. (MIRA 17:5)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut khimicheskikh  
reaktivov i osobo chistiykh khimicheskikh veshchestv.

LASTOVSKIY, R.P.; KOLPAKOVA, I.D.; IVANOVA, N.I.

m-Phenylenediamine-N,N,N',N'-tritraacetic acid. Met.  
poluch. khim. reak. i prepar. no.6:72-73 '62. (MIRA 17:5)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut khimicheskikh  
reaktivov i osobo chistykh khimicheskikh veshchestv.

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

BYKOV, A.M.; TOTSKIY, A.N.; KOLPAKOVA, L.D.

Vibratory burr removal from machine parts. Mashinostroitel'  
no.7:14-15 Jl '64. (MIRA 17:8)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

LASTOVSKIY, R.P.; KOLPAKOVA, I.D.

a,a'a"-Triaminebibenzyldiphenylmethane-N,N,N', N",N"-  
hexaacetic acid. Met. poluch. khim. reak. i prepar.  
no.6:73-74 '62. (MIRA 17:5)

I. Vsesoyuznyy nauchno-issledovatel'skiy institut khimicheskikh  
reaktivov i osobo chistykh khimicheskikh veshchestv.

*Kolpakova I. D.*

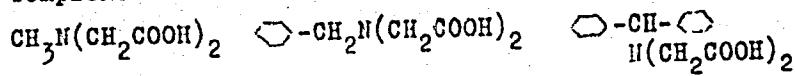
75-1-4/26

AUTHORS: Lastovskiy, R. P., Vaynshteyn, Yu. I.,  
Dyatlova, N. M., Kolpakova, I. D.

TITLE: New Complexones. (Novyye kompleksnye).  
Information 3. Benzylaminodiacetic Acid and  $\alpha, \alpha', \alpha''$ -Triaminodibenzylidiphenylmethanehexaacetic Acid  
(Soobshcheniye 3. Benzilamindiuksusnaya kislota i  $\alpha, \alpha', \alpha''$ -Triaminodibenzildifenilmetangeksauksusnaya kislota)

PERIODICAL: Zhurnal Analiticheskoy Khimii, 1958, Vol. 13, Nr 1,  
pp. 31-35 (USSR)

ABSTRACT: With the examples of methylaminodiacetic acid (1), benzylaminodiacetic acid (2) and benzhydrolaminodiacetic acid (3) the influence exerted by the modification of the molecular weight upon the complex-forming properties of some complexones was determined.



(1)

(2)

(3)

The investigation of the properties of these new compounds

~~Card 1/5~~

All-Union Sci Res. Inst. Chemical Reagents, Moscow

New Complexons .

75-1-4/26

Information 3. Benzylaminodiacetic Acid and  $\alpha,\alpha',\alpha''$ -Triaminodibenzyl-diphenylmethanehexaacetic Acid

was carried out polarographically. The displacement of the half-wave potentials for a number of cations at different  $p_H$  were also determined. In this connection it was found that benzylaminodiacetic acid at  $p_H$  2,5 forms complex compounds with the ions

$Cu^{2+}$ ,  $Bi^{3+}$ ,  $Ni^{2+}$  and  $Sb^{3+}$ , at  $p_H$  4,4 with the ions  $Cu^{2+}$ ,  $Co^{2+}$  and  $Mo(V)$ , at  $p_H$  9,35 with the ions  $Pb^{2+}$ ,  $La(III)$  and at  $p_H$  12,4 with the ions  $Cu^{2+}$ ,  $La(III)$  and  $Sb^{3+}$ .

A comparison between methylamine-, benzylamine- and benzhydrol amine-diacetic acid showed that an increase in molecular weight under certain conditions causes an increase in the complex-forming properties. The polarographic investigation of  $\alpha,\alpha',\alpha''$ -triaminodibenzyl-diphenylmethanehexaacetic acid (4) showed that this compound at  $p_H$  2,5 forms complex compounds with the ions

$Pb^{2+}$ ,  $Cu^{2+}$ ,  $As(III)$ ,  $Ni^{2+}$ ,  $Co^{2+}$  and  $Mo(VI)$ , at  $p_H$  4,4 with the ions  $Co^{2+}$ ,  $Mo(VI)$ ,  $Fe^{3+}$ , at  $p_H$  9,35 with the ions  $Pb^{2+}$ ,

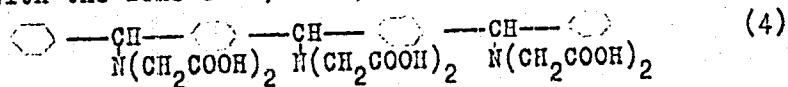
Card 2/5

75-1-4/26

## New Complexons..

Information 3. Benzylaminodiacetic Acid and  $\alpha, \alpha', \alpha''$ -  
 -Triaminodibenzylidiphenylmethanehexaacetic Acid

$\text{Bi}^{3+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Cr}^{3+}$  and  $\text{La(III)}$  and at  $p_{\text{H}}$  12,4  
 with the ions  $\text{Cu}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$  and  $\text{Al}^{3+}$ .



The formation of a number of complex compounds with this complexone is dependent on time. Thus, e.g., at  $p_{\text{H}} 9,35$  the half-wave potential of cadmium amounts to from -0,6 to -0,76 V, in this connection the height of the wave decreases from 16 to 11 mm and a second wave forms. The existence of two waves can here not be caused by a stepwise reduction, as cadmium does not show any intermediate stages in the oxidation number. The formation of two waves may be explained by the formation of different complex compounds so slowly passing over into one another that each of them is capable of forming its own wave. After 15 days standing the second wave disappears and the reduction potential of cadmium amounts to -0,7 V. On further standing no change

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New Complexons.

75-1-4/26

Information 3. Benzylaminodiacetic Acid and  $\alpha,\alpha',\alpha''$ -Triaminodibenzylidiphenylmethanehexaacetic Acid

any more occurs. This phenomenon may be explained by the presence of 3 complex-forming groups in  $\alpha,\alpha',\alpha''$ -triamine-dibenzylidiphenylmethanehexaacetic acid which form intermediary complexes which one after another enter into the reaction. For a more complete characterization of the investigated new complexones the dissociation constants of the formed complex compounds were determined in a polarographic way. For benzylaminodiacetic acid the dissociation constants of the complexes with copper and bismuth were determined, for the disodium salt of benzhydrolaminodiacetic acid the dissociation constants of the complexes with copper, cobalt, nickel, lanthanum and cadmium, and for  $\alpha,\alpha',\alpha''$ -triaminodibenzylidiphenylmethanehexaacetic acid the dissociation constants of the complexes with copper, lanthanum and cadmium. The results of the polarographic investigations of the disodium salt of benzhydrolaminodiacetic acid had already been published previously (ref. 1). The synthesis of benzylaminodiacetic acid and  $\alpha,\alpha',\alpha''$ -triaminodibenzylidiphenylmethanehexaacetic acid are accurately described. There are 2 tables, and 3 references, all of which are Slavic.

~~CARD 4/3~~

S/075/60/015/004/009/030/XX  
B020/B064

AUTHORS:

Lastovskiy, R. P., Kolpakova, I. D., and Dyatlova, N. M.

TITLE:

New Complexons. Information 4. Synthesis and Study of the  
Complexons of the Triazine Series

PERIODICAL:

Zhurnal analiticheskoy khimii, 1960, Vol. 15, No. 4,  
pp. 419 - 423

TEXT: Continuing their study of the synthesis of new complexons (Refs. 1-3), the authors investigate here the effect of nitrogen in the triazine cycle upon its capability of forming complex compounds. The introduction of atoms capable of coordinating with metals into the complexon molecule increases its capability of forming complexes and, in many cases, increases the selectivity of complexons for several metal cations. It was of interest to study the effect of heteroatoms in cyclic compounds. For this purpose, the following complexons containing a 1,3,5-triazine cycle were prepared: 2-oxy-4,6-diamino-1,3,5-triazine-  
N,N,N',N'-tetraacetic acid (I) and 2,4,6-triamino-1,3,5-triazine-  
N,N,N',N',N'',N''-hexaacetic acid (II) by condensing cyanur chloride with

Card 1/3

New Complexons. Information 4. Synthesis and Study of the Complexons of the Triazine Series S/075/60/015/004/009/030/XX  
B020/B064

imino diacetic acid. The complex-forming properties of the new compounds were polarographically studied by shifting the half-wave potential and determining the instability constants of the complexes of a number of cations. Table 1 indicates that the synthesized complexons form a number of compounds with metal ions, among which the following are of special interest: At pH 2.5, I reacts with  $Pb^{2+}$ ,  $Cu^{2+}$ ,  $Bi^{3+}$ ,  $Cd^{2+}$ ,  $Ni^{2+}$ ,  $Mo^{VI}$ , and  $Ti^{IV}$ ; at pH 4.4, apart from these ions, with  $As^{III}$  and  $Mn^{2+}$ ; at pH 9.35 with  $Pb^{2+}$ ,  $Cu^{2+}$ ,  $Cd^{2+}$ ,  $As^{III}$ ,  $Co^{2+}$ , and  $Mo^{VI}$ ; and at pH 12 with  $Cu^{2+}$ ,  $Cd^{2+}$ ,  $Zn^{2+}$ ,  $Ni^{2+}$ , and  $Bi^{3+}$ . At pH 4.4, II reacts with  $Pb^{2+}$ ,  $Cu^{2+}$ ,  $Mn^{2+}$ ,  $Mo^{VI}$ , and  $Ti^{IV}$ ; at pH 2.5, apart from these ions, with  $La^{III}$ ,  $Tl^{+}$ , and  $Zn^{2+}$ ; at pH 9.3 with  $Pb^{2+}$ ,  $Cu^{2+}$ ,  $As^{III}$ ,  $Mn^{2+}$ , and  $La^{III}$ ; and at pH 12 with  $Cu^{2+}$ ,  $Cd^{2+}$ ,  $Ni^{2+}$ , and  $Mo^{VI}$ . To determine the influence of nitrogen atoms in the hetero-cycle upon the stability of the complexes being formed, the properties of compounds I and II were compared with one another and with m-phenylene diamine-N,N,N',N'-tetraacetic acid, which were synthesized and

Card 2/3

New Complexons. Information 4. Synthesis and S/075/60/015/004/009/030/XX  
Study of the Complexons of the Triazine Series B020/B064

polarographically examined for the purpose. The instability constants of some complexes formed by the complexons examined with several metals were determined polarographically (Table 2). The half-wave potential shifts of the ion complexes with I and II are in all cases greater than with III, while the tendency toward forming stable complexes with I is greater than with II. I and II are characterized by the presence of the same group capable of forming complexes with metal cations, i.e.,  $(HOOC-CH_2)_2N-C-N=C-N(CH_2COOH)_2$ . The increased capability of II of forming complexes may be ascribed to the presence of a symmetrical molecule (three iminodiacetic acid groups). Finally, the synthesis of I and II is described in detail. There are 2 tables and 6 references: 4 Soviet, 1 Swiss, and 1 German.

ASSOCIATION: Vsescouznyy nauchno-issledovatel'skiy institut khimicheskikh reaktivov, Moskva (All-Union Scientific Research Institute for Chemical Reagents, Moscow)

SUBMITTED: April 14, 1959

Card 3/3

KOLPAKOVA, I. D.

Cand Chem Sci - (diss) "Synthesis of several new complexes and a study of their properties as a function of structure." Moscow, 1961. 7 pp; (Ministry of Higher and Secondary Specialist Education RSFSR, Moscow Order of Lenin Chemical Technology Inst imeni D. I. Mendeleyev); 150 copies; price not given; (KL, 6-61 sup, 198)

LASTOVSKIY, R.P. (Moscow, Bogorodskiy val.d.3); DYATLOVA, N.M. (Moscow, Bogorodskiy val.d.3); KOLPAKOVA, I.D. (Moscow, Bogorodskiy val.d.3); TEMKINA, V.Ya. (Moscow, Bogorodskiy val.d.3); LAVROVA, O.Yu. (Moscow, Bogorodskiy val.d.3)

New complexones and possibilities of their application in analytical chemistry. Acta chimica Hung 32 no.2:229-233  
'62;

1. Vsesoyuznyy nauchno-issledovatel'skiy institut khimicheskikh reaktivov.

KOLPAKOVA, L.L.

Changes in the plasma volume and protein composition of the blood serum in late pregnancy toxemias. Akush. i gin. no.1: 130-133 '65. (MIRA 18:10)

1. Kafedra akusherstva i ginekologii (zav... prof. A.M. Mazhibits) Novokuznetskogo instituta usovershenstvovaniya vrachey.

KOLPAKOVA, N.P.

Some problems in the theory of controlled systems. Priborostroenie  
no.1;9-11 Ja '64. (MIRA 17:2)

BPF(n)-2/BW(d)/BW(l) Pg-4/Pk-4/P1-l/P2-l/P3-l/Pu-l/Pee-2 IJP(c)  
AM/CIS

PR - AT5004116

S/0000/64/000/000 0144/0154

AKOVA, N. P.

j+1

Theory of systems with several regulatable coordinates

Prilozhnoye soveshchaniye po teorii invariantnosti i yeye primecheniyu v  
sistemakh, 2d, Kiev, 1962. Teoriya invariantnosti v sistemakh  
ego upravleniya (Theory of invariance in automatic control systems);  
Gostinyi, Moscow, Izd.-o Nauka, 1964, 144-151

control system, control theory, system stability, linear system,  
invariant system, connected system

The present work treats a variety of questions relating to the possibilities of a system which make its channels autonomous and invariant. In which forces impinge or improperly regulated coordinates or in which cross-connections are called "connected systems." Part 1 considers systems with connections made according to improperly regulated coordinates. Part 2 considers cross-connected, connected systems. In each case, the relationship between the system being autonomous and invariant is studied. Part 3 systems with connections according to regulations. Part 4 con-

L 34511-65

SP: AT5004116

ability properties of connected systems. Part 5 examines the de-  
the character of a dynamic connection on the properties of the re-  
sult and on the quality of the transition processes in each of the  
models. Part 6 considers the problem of the synthesis of systems with  
adjustable coordinates. Orig. art. has: 4 figures and 30 formulas.

None

14Sep64

ENCL: 00

SUB CODE: DP, IE

000

OTHER: 000

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

KOLPAKOVA, N.P., kand. tekhn. nauk

Choice of the structure of coupled control systems. Trudy MAI  
No. 155:21-56 '64. (MIRA 17:11)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

ROGATINA, Nina Prokof'yevna; POPOVA, Zinaida Fedorovna; ARTMANIS, Stella Andreyevna; MEL'NIKOVA, Nina Ivanovna; AVDEYEVA Yekaterina Semenovna; KUZNETSOVA, Irina Pavlovna; ZHEREBINA, Anna Semenovna; VOYEVODINA, Aleksandra Dmitriyevna; KOLPAKOVA, Ninel' Yevgrafovna; KHAYEVA, Aleksandra Afanas'yevna; DUNDUKOVA, Valentina Petrovna; LAUSTEN, A.G., nauch. red.; GABOVA, D.M., red.; VINOGRADOVA, G.A., tekhn. red.

[Women's and children's light dress] Zhenskoe i detskoe legkoe platk'e.  
Moskva, Gostekhizdat, 1962. 493 p. (MIRA 15:7)  
(Dressmaking)

KOLPAKOVA, O.V.

36866 . Kozhnyye sosudistyye reaktsii u bol'nykh s sindromom boley v oblasti serdtsa raelichnogo prciskhozhdeniya. Trudy Med. in-ta (Izhev. gos. med. in-t) t. IX, 1949, s. 233-41

SO: Letopis' Zhurnal' Nykh Statey, Vol. 50, Moskva, 1949

1. KOLPAKOVA, P. F.
2. USSR (600)
4. Fire Clay-Charysh Region
7. Geology and mineral resources of the eroded surface of the Charysh region in the Altai Territory. Izv. Glav. upr. geol. fon. no. 2 1947.
9. Monthly List of Russian Accessions, Library of Congress, March 1953. Unclassified.

BREGETOVA, N.G.; KOLPAKOVA, S.A.

Gamasoidea, parasites of the water rat (*Arvicola terrestris L.*) and inhabitants of its nests in the Volga delta. *Paraz.sbor.* 14:56-70 '52.  
(MLRA 6:6)

1. Zoologicheskiy institut Akademii Nauk SSSR, Saratovskiy institut "Mikrob." (Parasites--Water rats) (Volga Delta--Mites)

BEEGETOVA, N.G.; KOLPAHOVA, S.A.

Canasid mites (Parasitiformes, Gamaseidea), parasites of small  
murine rodents and inhabitants of their nests in the Volga Delta.  
Parazit. aber. 16:184-197 '56. (MIRA 9:7)

1. Zoolicheskiy institut Akademii nauk SSSR i Saratovskiy institut  
"Micreb".  
(Volga Delta--Mites) (Parasites--Rodentia)

KOLPAKOVA, S.A.

USSR/Zooparasitology - Acarina and Insect-Vectors of Disease  
Pathogens.

G-2

Abs Jour : Ref Zhur - Biol., No 5, 1958, 19670

Author : Kolpakova, S.A.

Inst :

Title : Ecology of Fleas Ceratophyllus (Nosopsyllus) Mokrzeckyi  
Wagn.

Orig Pub : Materialy k poznaniyu fauny i flory SSSR, Otd. zool.,  
1956, No 34(49), Ektoparazity, No 3, 149-161

Abstract : In the Volga bottom lands and delta and in the hillock  
sands 11,606 fleas were collected from 5567 rodents  
(chiefly house mice) and 1133 nests. On house and field  
mice, on common field mice, and in nests, C. mokrzeckyi  
predominated. The abundance of fleas on animals and in  
nests was increased in the fall (which corresponded  
with the growth of animal numbers), remained high until  
the beginning of spring, and markedly decreased in summer.

Card 1/2

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0

KOLEKOVA, T. A.

"Certain Conditions of Formation of Nuclei in Bacteria," Arkhiv biol. nauk  
(Archives of the Biological Sciences), 25, 1-3, 100-110, 1945

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824010006-0"

KOLPAKOVA, T.A.; GOLIYENBIYEVSKAYA, Z.I.; SHEVTSOVA, N.I.; RYEINA, M.I.;  
NIKITINA, N.N.; RYBAKOVA, L.F.; SHIPSHINA, N.D.; KORN, A.N.; KO-  
ROVKIN, B.F.; KOSYAKOV, K.S.; STEPNAIA, A.A.

Suggestions made at the September 29, 1963, conference of "La-  
boratornoe delo" readers, members of the Leningrad Society of Phy-  
sicians and Laboratorians. Lab. delo-10 no.4:256 '64. (MIRA 17:5)

1. Predsedatel' pravleniya Leningradskogo obshchestva vrachey-la-  
borantov (for Kolpakova).
2. Chleny pravleniya Leningradskogo ob-  
shchestva vrachey-laborantov (for all except Kolpakova).

KOLPAKOVA, T.A.; SUDAKOV, V.P.

Hydraulic turbopump units for machine irrigation.

Trudy TIIIMSKH no.8:3-15 '57. (MIRA 15:5)

(Soviet Central Asia-Irrigation)

(Pumping machinery)

S/064/60/000/007/008/010  
B020/B054

AUTHORS: Kolpakova, T. D. and Barannik, V. P.

TITLE: Improvement of the Properties of NB (PB) Corrosion Inhibitors

PERIODICAL: Khimicheskaya promyshlennost', 1960, No. 7, pp. 68 - 70

TEXT: The authors describe the shortcomings of the PB-5 inhibitor which consists of water-soluble condensation products of aniline and has a molecular weight of about 400-600; its most important shortcomings are the low stability of acid solutions of the inhibitor in the presence of  $Fe^{3+}$ , and its pure solubility in dilute HCl. Instead of aniline, the inhibitor NB-8 (PB-8) contains ethanol amine which is soluble in dilute acids, water, and lyes, but has a much lower protective action against strong HCl than the inhibitor PB-5. The authors investigated the properties of inhibitors formed with partial substitution of aniline by ethanol amine, i.e., which contained phenyl and ethanol groups alternatingly. They made nine preparations with aniline contents decreasing

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